

Formation of Functionalized Carbocycles via Base-Promoted Ring Opening/Brook Rearrangement/Allylic Alkylation of γ -Silyl- β,γ -epoxybutanenitrile Followed by Nitrile Anion Cyclization with Bis-Electrophiles

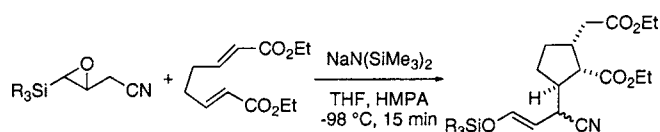
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ABSTRACT



Reaction of γ -silyl- β,γ -epoxybutanenitrile with a base generates an α -nitrile carbanion derivative of 4-siloxybut-3-enenitrile, which undergoes reaction with bis-electrophiles such as 1, ω -dihaloalkanes, ω -bromo- α,β -unsaturated esters, and bisenoates to provide highly functionalized carbocycles.

During our investigation of the mechanism of tandem base-promoted ring opening/Brook rearrangement/allylic alkylation of O -silyl cyanohydrins of β -silyl- α,β -epoxyaldehydes, we observed that treatment of **1** with NaN(SiMe₃)₂ (NHMDS)

in the presence of CH₃I afforded dimethylated products **3** in 32% yield in addition to monomethylated derivatives **2** (26%), suggesting that the second deprotonation and methylation are very fast processes, probably because of the small steric demand of the nitrile group (Scheme 1).²

This observation has prompted us to investigate reactions of **1** with a bis-electrophile such as **5**,^{3,4} which has two

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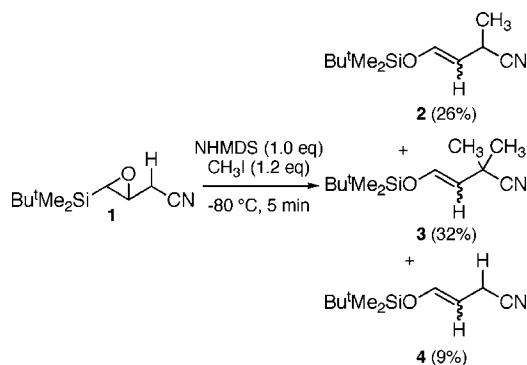
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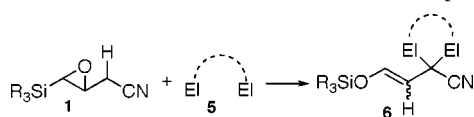
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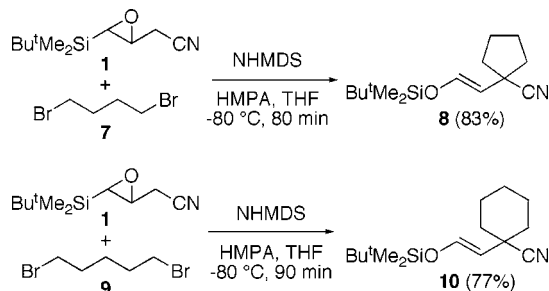
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Scheme 1. Base-Promoted Methylation of **1**

electrophilic sites in the molecule, allowing rapid access to highly functionalized carbocycles **6** (Scheme 2).

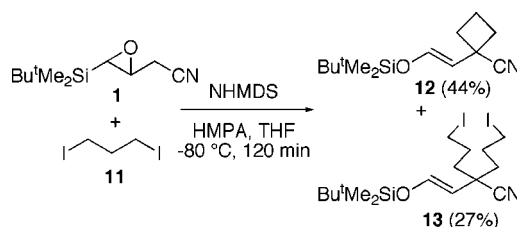
Scheme 2. Reaction of **1** with a Bis-Electrophile

We first examined the reaction of **1** with 1, ω -dihaloalkanes. When **1** in THF was treated with NHMDS (2.2 equiv) and HMPA (4.0 equiv) in the presence of 1,4-dibromobutane (**7**) at -80°C for 80 min, annulation product (*E*)-**8** was obtained in 83% yield (Scheme 3). Reaction

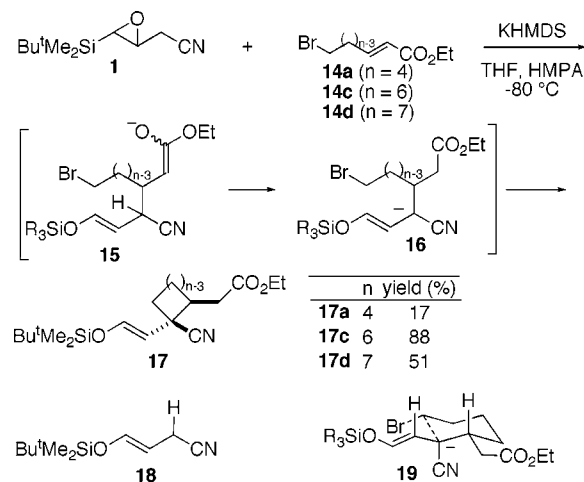
Scheme 3. Reaction of **1** with 1,4-Dibromobutane (**7**) and 1,5-Dibromopentane (**9**)

without HMPA or at elevated temperatures resulted in an increase in the formation of the (*Z*)-isomer. Similar results were obtained with **9** as the electrophile and for the four- and seven-membered rings that were synthesized, although uncyclized product **13** was formed as a byproduct with 1,3-diiodopropane (**11**) (Scheme 4).

Because the intramolecular alkylation of **1** proceeded well, we next examined the possibility that an annulation reaction using ω -bromo- α,β -unsaturated esters **14a–d**⁵ could occur

Scheme 4. Reaction of **1** with 1,3-Diiodopropane (**11**)

to provide cycloalkanecarboxylate derivatives **17** via a proton transfer in Michael adducts **15** followed by an intramolecular alkylation of **16** (Scheme 5).⁶ When epoxysilane **1** and **14a**

Scheme 5. Reaction of **1** with ω -Bromo- α,β -unsaturated Esters **14a,c,d**

were allowed to react with $\text{KN}(\text{SiMe}_3)_2$ (KHMDS) in THF at -80°C for 40 min, the expected annulation product **17a** was obtained in 17% yield together with unalkylated nitrile **18a** (73%). Use of other bases, including LDA and $\text{NaN}(\text{SiMe}_3)_2$ (NHMDS), increased the formation of **18** and decomposition products. Much better results were obtained with six- and seven-membered rings to give **17c** and **17d** in 88% and 51% yields, respectively. The stereochemistries of **17a** and **17c,d**, which were each obtained as a single diastereomer, were assigned on the basis of NOESY experiments. The selectivity observed in the case of **17a** was explained by assuming the least-hindered transition structure **19**.⁷

On the other hand, reaction with **14b** did not produce the corresponding five-membered ring. Instead, a Michael initiated ring closure (MIRC) reaction,^{8,9} involving conjugate

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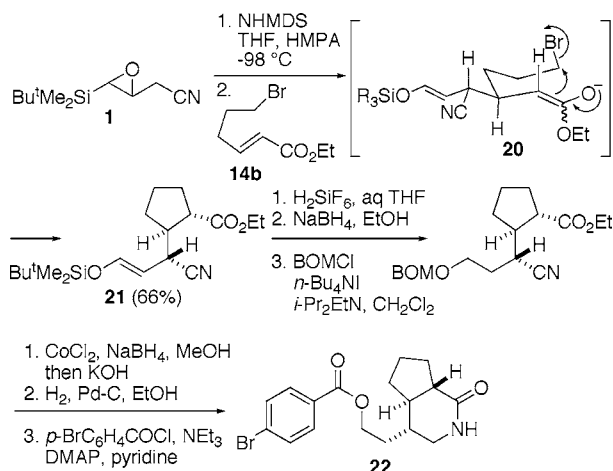
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addition to the enolate followed by an intramolecular alkylation, occurred to give **21** in 66% yield as a single diastereomer (Scheme 6). The stereochemical assignment was

Scheme 6. MIRC Reaction of **1** with **14b**

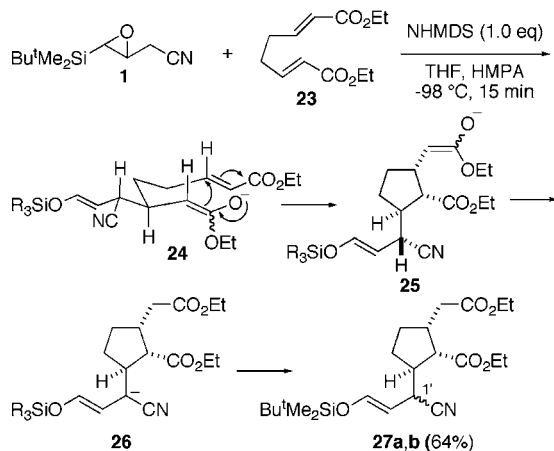


made on the basis of an X-ray analysis of **22**, which was derived from **21**.¹⁰ The observed stereochemical outcome can be understood by considering the transition structure **20** that minimizes unfavorable steric interactions.

The different course of the reaction depending on the ring size can be rationalized by assuming that the MIRC process (**20** → **21**) in the formation of a five-membered ring is faster than the intramolecular proton abstraction by the enolate anion in **15** ($n = 5$) (**15** → **16**) in comparison with that in the formation of four-, six-, and seven-membered rings. In this case, the best yield was obtained with NHMDS, and reaction with KHMDS and LDA resulted in a poor yield of **21** and an increase in the yield of **18**.

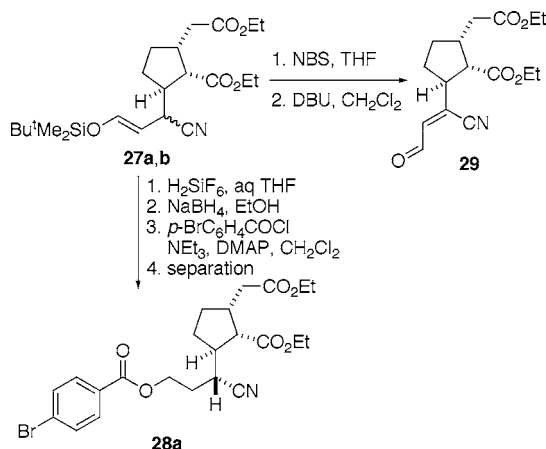
The above results led us to examine the MIRC reaction of **1** with α,ω -bisoates.¹¹ When a mixture of **1** and **23** in THF–HMPA was treated with NHMDS at $-98\text{ }^{\circ}\text{C}$ for 15 min, MIRC products **27a,b** were obtained as a diastereomeric mixture (8:3) in 64% yield (Scheme 7).

Scheme 7. MIRC Reaction of **1** with **23**



The stereostructures of **27a,b** were determined on the basis of X-ray analysis for **28a** (major isomer) that was derived from **27a,b** by a three-step sequence (Scheme 8) and from

Scheme 8. Derivatization of **27a,b**



the fact that a mixture of **27a** and **27b** upon treatment with NBS followed by DBU produced **29** as a single diastereomer, indicating that **27a** and **27b** were epimeric at C-1'. The observed stereoselectivity can be rationalized by assuming that the transition structure **24** is similar to **20**.^{11c} The fact that a mixture of diastereomers at C-1' was obtained in contrast to the case with **14b** above can be explained in terms of a proton transfer in **25** (**24** → **25**). This is supported by the finding that quenching the above reaction with MeI afforded a methylated product at the 1'-position. This MIRC reaction seems to be affected by a subtle conformational difference; in fact, reaction of **1** with α,ω -bisoates that would lead to a six-membered ring afforded only Michael addition products.

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Supporting Information Available: Full experimental details and spectral data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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